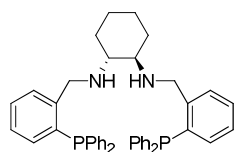
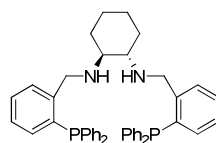




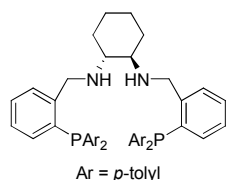
## Diaminodiphosphine Ligands



**K15-0026**  
**(1R,2R)-N,N-Bis(2-(diphenylphosphino)benzyl)cyclohexane-1,2-diamine, ≥97.0%**  
 $C_{44}H_{44}N_2P_2$ ; F.W: 662.78; [174758-63-5]

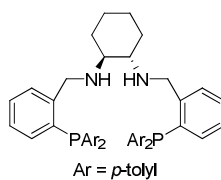


**K15-0027**  
**(1S,2S)-N,N-Bis(2-(diphenylphosphino)benzyl)cyclohexane-1,2-diamine, ≥97.0%**  
 $C_{44}H_{44}N_2P_2$ ; F.W: 662.78; [174677-83-9]



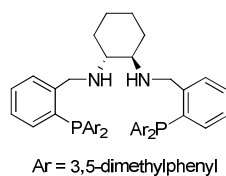
**K15-0088**  
**(1R,2R)-N,N-Bis(2-(di-p-tolylphosphino)benzyl)cyclohexane-1,2-diamine, ≥97.0%**  
 $C_{48}H_{52}N_2P_2$ ; F.W: 718.89

Ar = *p*-tolyl



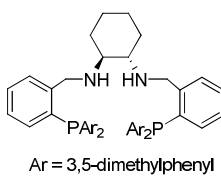
**K15-0089**  
**(1S,2S)-N,N-Bis(2-(di-p-tolylphosphino)benzyl)cyclohexane-1,2-diamine, ≥97.0%**  
 $C_{48}H_{52}N_2P_2$ ; F.W: 718.89

Ar = *p*-tolyl



**K15-0090**  
**(1R,2R)-N,N-Bis(2-(bis(3,5-dimethylphenyl)phosphino)benzyl)cyclohexane-1,2-diamine, ≥97.0%**  
 $C_{52}H_{60}N_2P_2$ ; F.W: 774.99

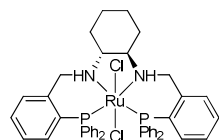
Ar = 3,5-dimethylphenyl



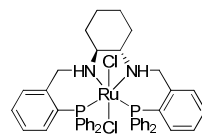
**K15-0091**  
**(1S,2S)-N,N-Bis(2-(bis(3,5-dimethylphenyl)phosphino)benzyl)cyclohexane-1,2-diamine, ≥97.0%**  
 $C_{52}H_{60}N_2P_2$ ; F.W: 774.99

Ar = 3,5-dimethylphenyl

## Ruthenium Diaminodiphosphine Catalysts



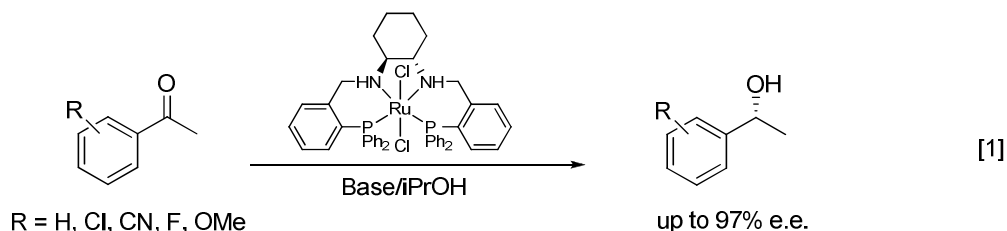
**K44-0042**  
**Dichloro((1R,2R)-N,N-Bis(2-(diphenylphosphino)benzyl)cyclohexane-1,2-diamine)ruthenium(II), ≥97.0%**  
 $C_{44}H_{44}Cl_2N_2P_2Ru$ ; F.W: 834.76; [429678-11-5]

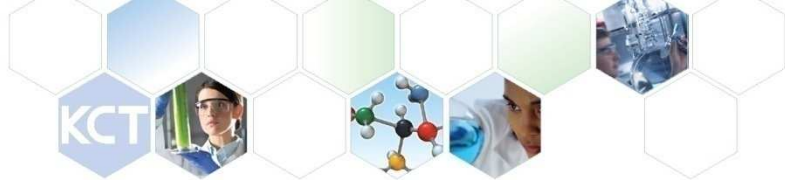


**K44-0043**  
**Dichloro((1S,2S)-N,N-Bis(2-(diphenylphosphino)benzyl)cyclohexane-1,2-diamine)ruthenium(II), ≥97.0%**  
 $C_{44}H_{44}Cl_2N_2P_2Ru$ ; F.W: 834.76; [302924-37-4]

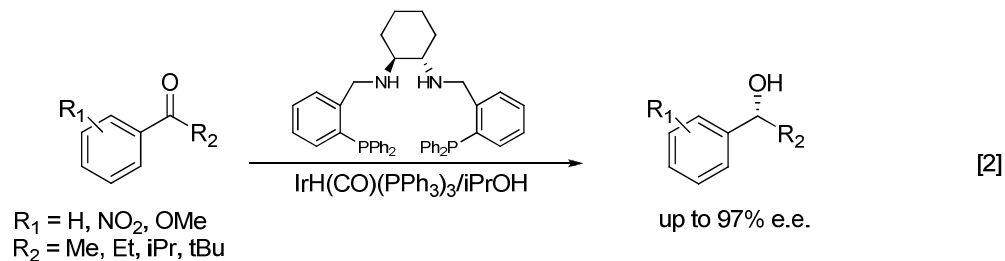
Kanata Chemical Technologies offers a series of chiral diaminodiphosphine ligands. The metal complexes of these ligands are excellent catalyst precursors for the asymmetric  $H_2$ <sup>1</sup> and transfer hydrogenation<sup>2-4</sup> of a series of ketones leading to the corresponding chiral alcohols with up to 99% e.e. (reactions 1-3). They are also effective catalysts for the enantioselective epoxidation of a number of unfunctionalized olefins with hydrogen peroxide,<sup>5</sup> and the kinetic resolution of racemic secondary alcohols.<sup>6</sup>

### Asymmetric Transfer Hydrogenation of Aromatic Ketones:<sup>2</sup>

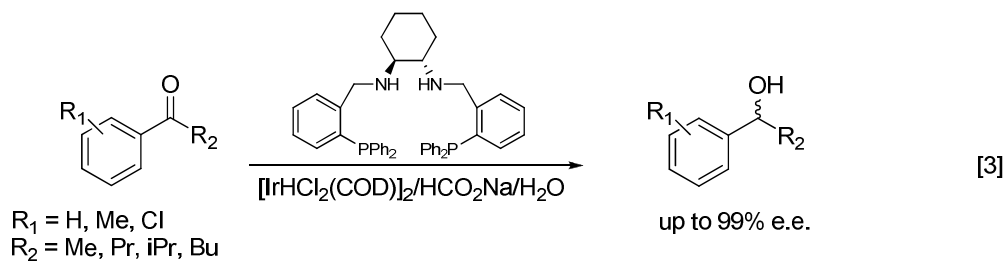




### Asymmetric Transfer Hydrogenation of Aromatic Ketones under Base-Free Conditions:<sup>3</sup>



### Asymmetric Transfer Hydrogenation of Aromatic Ketones in Aqueous Media:<sup>4</sup>



### References

1. Rautenstrauch *et al. Int. Pat. Appl.* WO2002/40155.
2. Gao, J. X. *et al. Organometallics* **1996**, *15*, 1087
3. Dong, Z. -R. *et al. Org. Lett.* **2005**, *7*, 1043.
4. Xing, Y. *et al. Tet. Lett.* **2006**, *47*, 4501.
5. Stoop, R. M. *et al. Organometallics* **2000**, *19*, 4117.
6. Li, Y. -Y. *et al. Org. Lett.* **2006**, *8*, 5565.